# Asymmetric Synthesis of (S)-Dimethyl -4, 4'-dimethoxy-5, 6, 5', 6'-dimethenedioxybiphenyl-2, 2'-dicarboxylate 

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#### Abstract

S)-Dimethyl-4, 4'-dimethoxy-5, 6, 5', 6'-dimethenedioxy-biphenyl-2, 2'-dicarbonylate was synthesized in reasonable yield through a series of reactions, including chiral oxazolinemediated asymmetric Ullmann coupling, from methyl 2-bromo-5-methoxy-3, 4-methenedioxybenzoate.


Keywords: Biphenyl, Schisandrin C, asymmetric Ullmann coupling.

Schizandra chinesis (wuweizi) has long been used in Chinese herbal medicine. Schisandrin C, as a constituent of Schizandra chinesis, shows various pharmacological activities ${ }^{1}$. The similar compound, dimethyl-4, 4'-dimethoxy-5, 6, 5', 6'-dimethenedi-oxy-biphenyl-2, $2^{\prime}$-dicarboxylate ( $\alpha$-DDB), discovered by $\mathrm{Xie}^{2,3}$ et al. in the investingation of schisandrin C, and some derivatives have also exhibited anti-HIV and anti-HBV activity ${ }^{4}$. Further study of this kind of biphenyls has attracted considerable attention.

It is well known that $2,2^{\prime}, 6,6^{\prime}$-tetra-substituents of biaryl make it difficult to rotate about aryl-aryl bond, which could result in two axially chiral isomers with R/S configuration. Furthermore, R and S isomers showed different biological activities ${ }^{5}$. To date, there is scarce report on the preparation of chiral $\alpha$-DDB besides classical resolution of $\alpha$-DDB's racemic isomers ${ }^{6}$. Therefore, it is necessary to find an efficient stereoselective method of synthesizing DDB. This paper described that S-DDB was prepared in reasonable yield.


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## Scheme 1



Reagent: (a) 1. KOH , in $95 \% \mathrm{EtOH}$; 2. Conc. HCl ; (b) $\mathrm{SOCl}_{2}$, in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; (c). (S)-2-aminopropanol, $\mathrm{Et}_{3} \mathrm{~N}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; (d) $\mathrm{SOCl}_{2}, \mathrm{Et}_{3} \mathrm{~N}$; (e) $t$-ButOK, THF and $t$-ButOH; (f) activated copper dust in DMF; (g) 1. TFA, water. 2. $\mathrm{Ac}_{2} \mathrm{O}$, DMAP; (h) NaOMe , in MeOH .
(S)-DDB was synthesized as Scheme1 from methyl 2-bromo-5-methoxy-3, 4-methenedioxybenzoate $\mathbf{1}$ via a series of reactions including the key chiral oxazolinemediated asymmetric Ullmann coupling reaction ${ }^{7}$.
$\mathbf{1}$ was hydrolyzed to afford $\mathbf{2}$, and then $\mathbf{2}$ was converted to the carboxylic chloride $\mathbf{3}$ by the reaction with $\mathrm{SOCl}_{2}$. 3 was mixed with ( S )-2-amino propanol and triethylamine to afford 4. Reaction of the amide 4 with $\mathrm{SOCl}_{2}$ gave 5 , and 5 was heated in the presence of potassium $t$-butoxide in the mixed solvent of THF and $t$-butanol to give (S)-oxazoline phenyl bromide 6.

The asymmetric Ullmann coupling reaction of $\mathbf{6}$ in the presence of activated copper powder gave the mixture of bis-oxazoline biphenyl 7S and 7R. HPLC and ${ }^{13} \mathrm{C}$ NMR analysis indicated that the diastereoisomeric ratio of S and R was 81:19. Recrystallization in the mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and light petroleum afforded the optical pure 7S.

S-Configuration of the biphenyl subunit of 7S was confirmed by Cotton effect in CD spectrum. The CD spectroscopy of 7S exhibited negative Cotton effect at 305 nm , 260 nm and positive Cotton effect at 293 nm in methanol. Therefore, the configuration of compound 7S was assigned as $\mathrm{S}^{8}$. In addition, the configuration was also confirmed by the specific rotation of the related compound S-DDB.

As a pair of diastereoisomers, 7R and 7S could be separated on HPLC by general C 18 column. The retention time of 7S and 7R were 25.7 min and 27.6 min , respectively, at given condition. In addition, ${ }^{13} \mathrm{CNMR}$ spectrum of 7 S showed twelve single-peaks, but the mixture of 7S with minor 7R exhibited twelve pairs of double-peaks. The peak intensity ratio of double-peaks was approximately consistent with the ratio of S and R measured by HPLC.

7S was hydrolyzed and reacted with acetic anhydride to gave diamide $\mathbf{8}$, and the crude of $\mathbf{8}$ was alcoholysized by methanol in the presence of sodium methoxide to afford S -DDB. The CD spectroscopy showed negative Cotton effect at 259 nm . Therefore, the configuration of $\mathbf{8}$ was assigned as $\mathrm{S}^{9}$.

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## Experimental

${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Bruker-300 MHz. IR spectra, ESI-MS and circular dichroism spectroscopy were carried out on a Shimadazuir-408, Bruke Esquire-3000 and Jasco J-20, respectively. HPLC were on Shimadzu LC-10A.

Compound 4: Methyl 2-bromo-5-methoxy-3,4-methenedioxybenzoate $\mathbf{1}(6 \mathrm{~g}, 20.7$ mmol ) was added to water ( 60 mL ) with $\mathrm{KOH}(4.5 \mathrm{~g}, 69 \mathrm{mmol})$. The mixture was refluxed for 7 h , and then acidified with con. HCl till $\mathrm{pH}=3.0$. Filtration and dryness of the mixture gave corresponding benzoic acid $2(5.48 \mathrm{~g}, 96 \%) . \mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL}), \mathrm{SOCl}_{2}$ ( $2.8 \mathrm{~mL}, 39 \mathrm{mmol}$ ) and 2 drops DMF were added to compound $2(5.20 \mathrm{~g}, 18.9 \mathrm{mmol}$ ). After the mixture was stirred for 12 h , the solution was evaporated under reduced pressure. Toluene ( 10 mL ) was added to the residue. The solvent was evaporated again to afford crude 3 . The solution of $\mathbf{3}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ was added to the mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$, (S)-2-amino propanol( $1.56 \mathrm{~g}, 20.8 \mathrm{mmol}$ ) and triethyl- amine $(5 \mathrm{~mL})$ at cooling in ice bath. This mixture was stirred for 12 h at r.t. under Ar, and poured into 100 mL water. The organic layer was separated and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and evaporated in vacuo. Recrystallization from ethanol gave $4(5.27 \mathrm{~g})$ in $84 \%$ yield. mp. $175-176{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta \mathrm{ppm}$ ) 6.97(s, $\left.1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right), 6.10\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{O}\right), 4.28(\mathrm{~m}, 1 \mathrm{H}$, NCH), 3.92(s, 3H, OCH ${ }_{3}$ ), 3.83-3.78(dd, 1H, J=3.3, $10.8 \mathrm{~Hz}, \mathrm{OCH}$ ), 3.69-3.64(dd, 1H, $\mathrm{J}=5.4,10.8 \mathrm{~Hz}, \mathrm{OCH}), 1.31\left(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.9, \mathrm{CH}_{3}\right) . \operatorname{IR}\left(\mathrm{KBr}, \mathrm{cm}^{-1}\right) 3278.8(\mathrm{O}-\mathrm{H}), 1643.2$, 1624.0 (CONH). ESI-MS $(\mathrm{m} / \mathrm{z}) 332$ [M+H] ${ }^{+}$.

Compound 6: $\mathrm{SOCl}_{2}(2.40 \mathrm{~mL}, 33 \mathrm{mmol})$ was added to the solution of $\mathbf{4}(5 \mathrm{~g}, 15$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$. The mixture was stirred at r.t. overnight and poured into 120 mL ice water. The aqueous $10 \% \mathrm{NaOH}$ was added till pH 7 . The organic layer was separated and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and the solvent was removed in vacuo to give 5 . To 5 were added THF ( 75 mL ), potassium $t$-butoxide ( $3.36 \mathrm{~g}, 30 \mathrm{mmol}$ ) in $t$-butanol( 75 mL ). The mixture was stirred at reflux overnight. The solvent was removed, 30 mL water was added and then was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 50 \mathrm{~mL})$. The $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ extract was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated. The residue was purified by silica gel chromatography to yield (S)-oxazoline phenyl bromide 6 in 95\% yield ( 4.5 g ). mp.83-84 ${ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}{ }^{25}=-36.1\left(\mathrm{c} 3.1, \mathrm{CHCl}_{3}\right),{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta \mathrm{ppm}$ ) 7.06(s, $\left.1 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right)$, $6.10\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{O}\right), 4.54-4.48\left(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}=7.8,9 \mathrm{~Hz}, \mathrm{OCH}^{\mathrm{a}}\right), 4.43-4.38(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH})$, $3.97\left(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}, \mathrm{OCH}^{\mathrm{b}}\right), 3.91\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 1.38\left(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6.3 \mathrm{~Hz}, \mathrm{CH}_{3}\right)$. ESI-MS $(\mathrm{m} / \mathrm{z}) 314[\mathrm{M}+\mathrm{H}]^{+}, 336[\mathrm{M}+\mathrm{Na}]^{+}$.

Compound 7S: To the anhydrous DMF(10 mL) was added (S)-oxazoline phenyl bromide $\mathbf{6}(1.5 \mathrm{~g}, 4.77 \mathrm{mmol})$, activated copper powder( $1.5 \mathrm{~g}, 23.5 \mathrm{mmol})$. The mixture was heated under Ar at $145^{\circ} \mathrm{C}$ for 24 h , and the solvent was removed in vacuo. To the residue was added $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ and washed with aqueous ammonia( $5 \%, 2 \times 15 \mathrm{~mL}$ ). The organic layer was separated and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated in vacuo. The residue was purified through silica gel chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ : $\left.\mathrm{EtOAC}: T H F=5: 2: 1\right)$ to give the mixture of 7S with minor 7R 1.04 g (93\%) in 62\%de (HPLC, C18-column, $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}=55 / 45,1 \mathrm{~mL} / \mathrm{min}, 7 \mathrm{~S}: \mathrm{Rt}=25.7 \mathrm{~min}$, 7 R : $\mathrm{Rt}=27.6 \mathrm{~min}$ ). Recrystallization from the mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and light petroleum afforded 7 S in $99 \% \mathrm{de}(0.513 \mathrm{~g}, 46 \%)$. mp. 162-163 ${ }^{\circ} \mathrm{C} .[\alpha]_{\mathrm{D}}^{25}-107(\mathrm{c} 0.036, \mathrm{EtOH}),{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta \mathrm{ppm}$ ): 7.23(s,
$2 \mathrm{H}, \mathrm{Ar}-\mathrm{H}), 5.99\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=1.2 \mathrm{~Hz}, \mathrm{OCH}_{2} \mathrm{O}\right), 5.95\left(\mathrm{~d}, 2 \mathrm{H}, \mathrm{J}=1.2 \mathrm{~Hz}, \mathrm{OCH}_{2} \mathrm{O}\right), 4.25-4.15(\mathrm{~m}$, 4 H , oxazoline $\mathrm{OCH}_{2}$ ), 3.95(s, 6H, $\mathrm{OCH}_{3}$ ), 3.77-3.71(m, 2H, NCH), $1.20(\mathrm{~d}, 6 \mathrm{H}, \mathrm{J}=3.2 \mathrm{~Hz}$, $\mathrm{CH}_{3}$ ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 163, 147, 142, 136, 121, 111, 109, 101, 73, 61, 56, 21; ESI-MS $(\mathrm{m} / \mathrm{z}) 469[\mathrm{M}+\mathrm{H}]^{+}$. The CD spectroscopy showed the negative Cotton effect at $305 \mathrm{~nm}, 260 \mathrm{~nm}$ and positive Cotton effect at 293 nm in methanol.
(S)-DDB: To the solution of trifluoroacetic acid ( $0.20 \mathrm{~mL}, 2.50 \mathrm{mmol}$ ) in THF ( 15 $\mathrm{mL})$ and water $(0.5 \mathrm{~mL})$ was added compound $7 \mathrm{~S}(0.40 \mathrm{~g}, 0.85 \mathrm{mmol})$. The mixture was stirred at r.t. overnight, and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solvent was removed in vacuo. To the residue was added $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$, pyridine $(2 \mathrm{~mL})$, acetic anhydride $(0.5 \mathrm{~mL}$, 5.2 $\mathrm{mmol})$ and 4 -dimethylaminopyridine $(5 \mathrm{mg}, 0.04 \mathrm{mmol})$. The mixture was stirred at r.t. overnight. To the mixture was added methanol ( 5 mL ), stirred at r.t. for 5 h , washed with $4 \% \mathrm{HCl}(3 \times 25 \mathrm{~mL})$ and water ( 25 mL ) and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solvent was removed in vacuo to obtain crude 8. To the crude $\mathbf{8}$ was added methanol ( 10 mL ), $\operatorname{THF}(10 \mathrm{~mL})$ and $\mathrm{NaOMe}(0.25 \mathrm{~g}, 4.5 \mathrm{mmol})$, and stirred at r.t. overnight. The solvent was removed in vacuo. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ and washed with water. After the organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated in vacuo, the residue was purified by silica gel column chromatography to afford (S)-DDB ( 0.26 g , $73 \%$ ) in 93\%ee (Chiral HPLC analysis: chiralcel OD column, ethanol/hexane=15/85, 0.5 $\mathrm{mL} / \mathrm{min}$ ). The CD spectroscopy of (S)-DDB exhibited negative Cotton effect at 259 nm . $[\alpha]_{\mathrm{D}}{ }^{25}=-76.32\left(\mathrm{c} 0.41, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) . \mathrm{mp} .140-142{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \delta \mathrm{ppm}\right)$ 7.42 (s, 2H, 2Ar-H), 6.06(s, 4H, $2 \mathrm{CH}_{2}$ ), 4.02(s, $6 \mathrm{H}, 2 \mathrm{Ar}-\mathrm{OCH}_{3}$ ), $3.78\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{COOCH}_{3}\right)$. ESI-MS( $\mathrm{m} / \mathrm{z}$ ) $419[\mathrm{M}+\mathrm{H}]^{+}$.

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